

Q.8 Explain the Terms.

Ans. Chimney Efficiency: As the height of chimney depends upon temp. of hot gases leaving the chimney. This means that natural or chimney draught created at the cost of thermal efficiency of the boiler plant. In artificial draught the flue gases can be made to leave chimney at a reduced temp. and thus amount of heat being carried away by the flue gases would be less in this case.

Let  $T =$  temp. of flue gases,  $T_2 =$  temp. of flue gases in artificial draught. The extra heat carried away by one kg. of flue gas due to higher temp. required to produce the natural draught.

$$= 1 \times C_p(T - T_2) \text{ kJ}$$

Natural draught in terms of column of hot gases.

$$= H \left[ \left( \frac{M}{m+1} \right) \times \frac{T}{T_1} - 1 \right] m.$$

The efficiency of chimney is

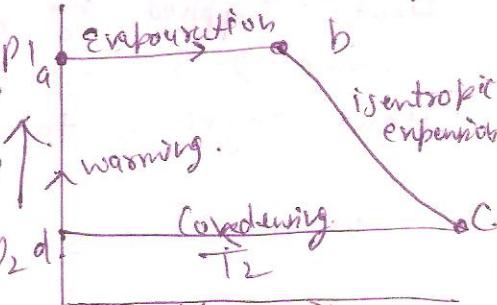
$$\eta = \frac{H \left[ \left( \frac{M}{m+1} \right) \times \frac{T}{T_1} - 1 \right]}{C_p(T - T_2)} \times 100\%$$

Q.9 Discuss the effect of operating condition on the performance of Rankine cycle.

Ans. Rankine cycle is the ideal cycle on which a steam engine works. Let us consider one kg. of water at a temp. of  $T_2$  and under a pressure of  $P_2$  bar absolute.

(i) Operation  $a-b$ : A hot body is supplied to this one kg. of water in the cylinder such that its pressure and temp. are raised from  $P_2$ ,  $T_2$  to  $P_1$  and  $T_1$ .

(ii) Operation  $b-c$ : Evaporation of water takes place at const. pressure. A certain amount of latent is gained from the hot body.



(iii) Operation  $b-c$ : At  $b$  the hot body is removed while the steam is allowed to expand isentropically to the pressure  $P_2$  and volume  $V_2$ . The temp. also falls to  $T_2$ . A certain amount of work is done on the piston.

(iv) Operation  $c-d$ : Heat is rejected by the steam to the cold body brought in contact with it. The heat is rejected at constant pressure and the steam gets condensed into water. Thus the cycle is completed.

Efficiency of Rankine cycle:

$$\begin{aligned} \text{Total heat absorbed} &= \text{heat absorbed during } da + \text{heat absorbed during } ab \\ &= (h_{f1} - h_{f2}) + n_1 h_{fg} \end{aligned}$$