

Q19
 Ans. The volumetric efficiency of a single stage reciprocating compressor is defined as the volume of gas entering the compressor or per min. divide by the piston displacement of the compressor per minute. Therefore for a single stage compressor.

$$\text{Volumetric efficiency} = \frac{\text{Vol. of gas entering compressor per minute}}{\text{Piston displacement per minute.}}$$

The volumetric eff. of multistage compressor or is defined in the same way except that the piston displacement taken by that of the low pre. cylinder only. therefore for a single stage compressor.

$$\text{Volumetric compressor} = \frac{\text{Vol. of gas entering compressor per minute}}{\text{Piston disp. of the low pre. cylinder per minute}}$$

As by the fig.

$$\text{Volumetric } \eta = \frac{V_1 - V_4}{V_1 - V_3}$$

from the process 3-4

$$\left(\frac{P_3}{P_4}\right)^{1/n} = \frac{V_4}{V_3} = \left(\frac{P_2}{P_1}\right)^{1/n} \text{ or } V_4 = V_3 \times \left(\frac{P_2}{P_1}\right)^{1/n}$$

Since $P_3 = P_2$ & $P_4 = P_1$

$$\text{Let the clearance ratio} = \frac{\text{clearance vol.}}{\text{swept vol.}} = \frac{V_3}{V_1 - V_3} = k.$$

$$\text{then volumetric efficiency} = \frac{V_1 - V_4}{V_1 - V_3} = \frac{V_1 - V_3 \left(\frac{P_2}{P_1}\right)^{1/n}}{V_1 - V_3} = \frac{V_1 - V_3 \left(\frac{P_2}{P_1}\right)^{1/n}}{V_3/k}$$

$$= \frac{kV_1}{V_3} - k \left(\frac{P_2}{P_1}\right)^{1/n}$$

$$\text{Since } k = \frac{V_3}{V_1 - V_3} \text{ so } V_1 = \frac{kV_3}{k} + V_3 = V_3 \left(\frac{1}{k} + 1\right)$$

$$\therefore \text{Volumetric efficiency} = k \frac{V_3 \left(\frac{1}{k} + 1\right)}{V_3} - k \left(\frac{P_2}{P_1}\right)^{1/n}$$

$$= 1 + k - k \left(\frac{P_2}{P_1}\right)^{1/n} = 1 + k - k \left(\frac{V_1}{V_2}\right)^{1/n}$$

hence if $P_2 = P_1$ the volumetric eff. will be 100% whereas if $V_1 = V_2$ the volumetric η will be zero.

$$= 1 + k - k \left(\frac{P_2}{P_1}\right)^{1/n} \times \frac{P_1 T_1}{P_2 T_2}$$

$$= 1 + k - k \left(\frac{V_1}{V_2}\right) \times \frac{P_1 T_1}{P_2 T_2}$$

where suffix 1 & 2 stand for inside & atm. conditions.

Q20

Ans. As we know
 mass of steam (kg) discharged through the nozzle per second

$$m = \frac{A}{v_1} \times \left[2 \left(\frac{n}{n-1}\right) P_1 v_1 \left\{ \left(\frac{P_2}{P_1}\right)^{2/n} - \left(\frac{P_2}{P_1}\right)^{(n+1)/n} \right\} \right]$$

where $h = 1.135$ for dry saturated steam

- $P_1 =$ pre. of steam at entry in N/m^2 absolute.
- $V_1 =$ vol. of one kg. of steam at pre. P_1 in m^3
- $P_2 =$ pre. of steam N/m^2 abs.
- $A =$ Area of cross section (m^2) of nozzle at the throat.
- $V_2 =$ vol. of one kg. of steam at absolute pre. P_2

Now there is only one value of the ratio (P_2) to which the discharge of steam through the nozzle is maxm. The value (P_2/P_1) is called critical pre. ratio. Therefore we differentiate only that portion which is inside the square brackets & equate it to zero.

$$i.e. \frac{2}{n} \left(\frac{P_2}{P_1}\right)^{2/n-1} - \frac{h+1}{n} \left(\frac{P_2}{P_1}\right)^{h/n-1} = 0 \text{ or } \frac{2}{n} \left(\frac{P_2}{P_1}\right)^{\frac{2-h}{n}-\frac{h+1}{n}} = 0$$

$$= \left(\frac{P_2}{P_1}\right)^{\frac{2-h}{n}-\frac{h+1}{n}} = \frac{h+1}{n} \times \frac{n}{2} = \frac{h+1}{2}$$

$$\Rightarrow \left(\frac{P_2}{P_1}\right)^{\frac{1-h}{n}} = \frac{h+1}{2} = \left(\frac{2}{h+1}\right)^{1-h} \text{ or } \frac{P_2}{P_1} = \left(\frac{2}{h+1}\right)^{\frac{h}{1-h}}$$

$$\Rightarrow \left(\frac{P_2}{P_1}\right) = \left(\frac{2}{h+1}\right)^{\frac{h}{h-1}} \Rightarrow \frac{P_2}{P_1} = \left(\frac{2}{h+1}\right)^{\frac{h}{h-1}}$$

now for dry saturated is 1.135 & critical pre. ratio is 0.578.

$\frac{h_2}{h_1}$ It may be defined as the ratio of isentropic head drop in the moving blades to isentropic head drop at the reaction turbine.

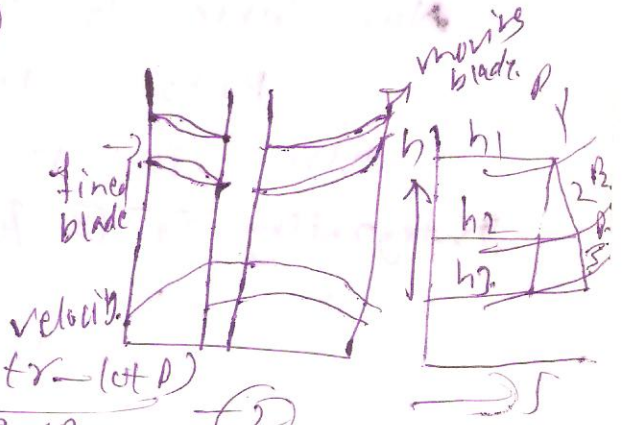
Two degree of reaction (R) = $\frac{\text{Enthalpy drop in moving blades}}{\text{Enthalpy drop in stage}} = \frac{h_2 - h_3}{h_1 - h_3}$

= $\frac{\text{Increase in the relative K.E in moving blade}}{\text{Stage work output}}$

$$\frac{V_{r0}^2 - V_{r1}^2}{V_b(V_{w1} - V_{w0})} = \frac{V_{v0}^2 - V_{v1}^2}{2 V_b (V_{w1} - V_{w0})}$$

$V_{r0} = V_f \cos \alpha$ & $V_{r1} = V_f \cos \beta$
 $V_{w1} + V_{w0} = V_f (\cot \beta + \cot \alpha)$
 $V_f =$ vel. of flow which remains const. through the stages

$$\therefore R = \frac{V_f^2 (\cos^2 \alpha - V_f^2 \cos^2 \beta)}{2 V_f (\cot \beta + \cot \alpha) V_b} = \frac{V_f (\cot \alpha - \cot \beta)}{2 V_b}$$



For 50% degree of reaction $V_b = V_f (\cot \alpha - \cot \beta)$

from the vel. triangle V_b can also be expressed as under

$$V_b = V_f (\cot \alpha - \cot \beta) = V_f (\cot \alpha - \cot \beta)$$